

Asymmetric Alcohol C–H Allylation and *syn*-Crotylation: C9–C20 of Tetra fibrin

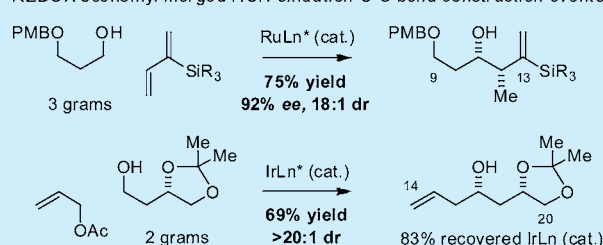
Takahiko Itoh, T. Patrick Montgomery, Antonio Recio, III, and Michael J. Krische*

University of Texas at Austin, Department of Chemistry and Biochemistry, Austin, Texas 78712, United States

Supporting Information

ABSTRACT: The C9–C20 segment of the fibrinogen receptor inhibitor tetra fibrin was prepared in 10 steps (longest linear sequence). Ruthenium catalyzed enantioselective *syn*-crotylation is used to construct C9–C13. Iridium catalyzed asymmetric alcohol C–H allylation of a commercial malic acid derived alcohol is used to construct C14–C20. Recovery and recycling of the iridium catalyst is described.

REDOX economy: merged ROH oxidation–C–C bond construction events.



Tetra fibrin¹ is a potent fibrinogen receptor inhibitor isolated from *Streptomyces neyagawaensis* NR0577 that functions by blocking GPIIb/IIIa receptors on the platelet surface, thus inhibiting platelet aggregation.^{2,3} While biosynthetically related to the oxo-polyene macrolide antibiotic lienomycin,⁴ tetra fibrin is acyclic and inactive against *Bacillus subtilis* and *Escherichia coli*. Further, the structure of tetra fibrin, which includes alternating 1,3-diol and 1,5-enediol motifs, represents a significant departure from all known fibrinogen receptor antagonists. Tetra fibrin has garnered interest with regard to the study of fibrinogen binding, platelet aggregation, and as a treatment for arterial thrombosis,^{3,5} which in turn has evoked interest in its preparation *via* total synthesis. Despite efforts reported from the laboratories of Cossy,⁶ Curran,⁷ Friestad,⁸ Roush,⁹ and the present author,¹⁰ including syntheses of a protected derivative of the natural product by Curran^{7d} and of *N*-acetyl dihydrotetra fibrin methyl ester by Roush,^{9c} the total synthesis of tetra fibrin remains an unmet challenge.

Using catalytic C–C bond-forming processes that occur through the addition, transfer, or removal of hydrogen,¹¹ total syntheses of 6-deoxyerythronolide B,^{12a} bryostatin 7,^{12b} trienomycins A and F,^{12c} cyanolide A,^{12d} and roxaticin^{12e} were completed in our laboratory. Due to the redox economy¹³ of the methods underlying their construction, the aforementioned syntheses represent the most concise routes to any member of their respective natural product families.^{11d} Inspired by the unanswered synthetic challenges posed by tetra fibrin, a 12 step route to the C21–C40 segment was devised using iridium catalyzed alcohol C–H allylation developed in our laboratory.^{10,14} Here, we report a 10 step synthesis of the C9–C20 substructure of tetra fibrin featuring enantioselective methods for ruthenium catalyzed alcohol C–H *syn*-crotylation¹⁵ and iridium catalyzed alcohol C–H allylation.¹⁴

The C9–C20 substructure of tetra fibrin was envisioned to arise through the reductive coupling of bromide 6 to epoxide 13 (Scheme 1). Synthesis of bromide 6 was readily

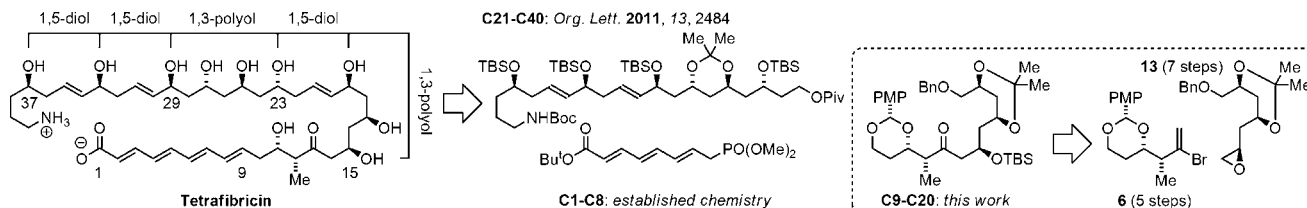
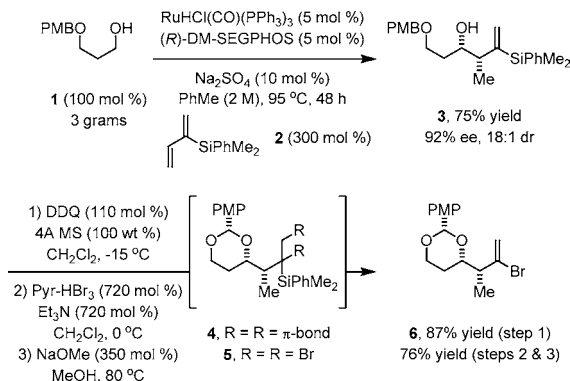
accomplished through ruthenium catalyzed hydrohydroxyalkylation of 2-silyl-butadiene 2 with the monoprotected 1,3-propanediol 1. A modification of our previously reported conditions,^{15a} which involves use of the catalyst generated from RuHCl(CO)(PPh₃)₃ and (*R*)-DM-SEGPHOS in the presence of substoichiometric sodium sulfate, provided the desired product of diene hydrohydroxyalkylation 3 in 75% yield with excellent control of *syn*-diastereo- and enantioselectivity (92% ee, 18:1 dr). Adduct 3 was converted to the *p*-methoxy benzylidene acetal in 87% yield upon DDQ oxidation.¹⁶ Finally, exposure of the alkenyl silane to pyridinium bromide perbromide gave the crude dibromide 5, which was treated with sodium methoxide to furnish the alkenyl bromide 6 (Scheme 2).¹⁷

Synthesis of the C14–C20 epoxide 13 began with the asymmetric iridium catalyzed alcohol C–H allylation¹⁴ of the commercially available chiral alcohol 7 derived from (*S*)-malic acid. Using the cyclometalated iridium catalyst “(S)-I” generated *in situ* from [Ir(cod)Cl]₂, 4-chloro-3-nitrobenzoic acid, allyl acetate, and (*S*)-BINAP, the desired homoallylic alcohol 8 was obtained in 69% isolated yield as a single diastereomer as determined by ¹H NMR of the crude reaction product. As previously noted,¹⁸ cyclometalated iridium catalysts of this type are remarkably stable to conventional silica gel flash chromatography. Given evidence that the carbonyl addition is turnover limiting,^{14b} the chromatographically stable π -allyl complex “(S)-I” is a potential catalyst resting state. In accord with this interpretation, “(S)-I” can be recovered from the reaction mixture and recycled to provide homoallylic alcohol 8 in 53% yield with equally high levels of diastereoselectivity. No further erosion in yield or selectivity was observed upon further iterations of recovery and recycling. Catalyst recovery and recycling, along with the ability to bypass discrete alcohol-to-

Received: December 9, 2013

Published: January 14, 2014

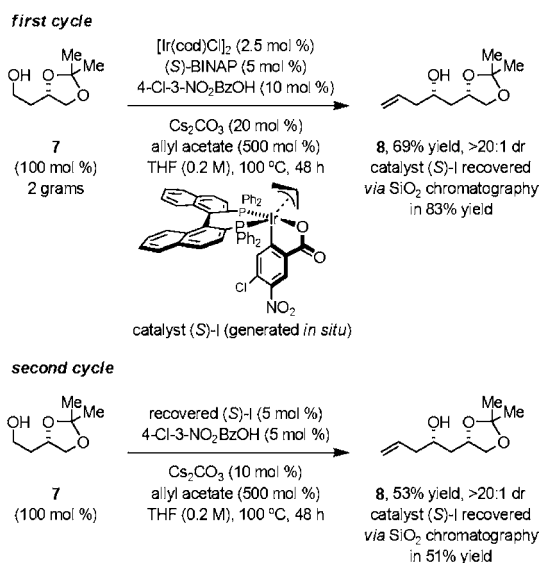
Scheme 1. Retrosynthetic Analysis of the Fibrinogen Receptor Inhibitor Tetrafibricin

Scheme 2. Synthesis of Alkenyl Bromide 6 via Ruthenium Catalyzed Hydrohydroxyalkylation of Diene 2^a

^aYields are of material isolated by silica gel chromatography. (R)-DM-SEGPHOS: (R)-(+)-5,5-bis(diphenylphosphino)-4,4-bi-1,3-benzodioxole. See Supporting Information for further details.

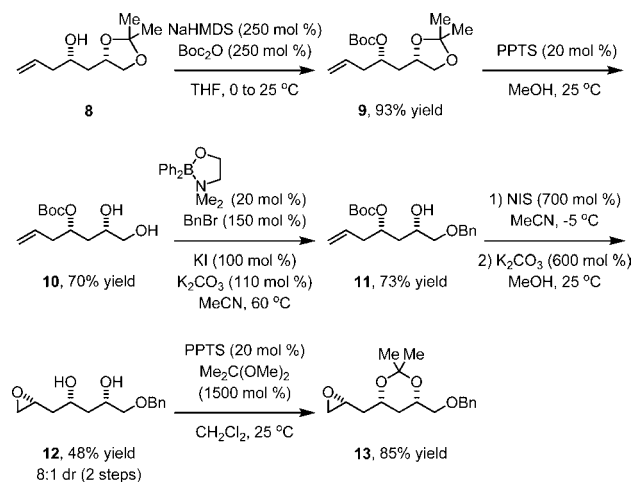
aldehyde redox reactions, renders our asymmetric allylation protocol more cost-effective (Scheme 3).

To complete the synthesis of epoxide 13, homoallylic alcohol 8 was converted to the *tert*-butyl carbonate 9, which was subjected to conditions for acid catalyzed methanolysis to furnish diol 10. Primary benzyl protection of diol 10 using Taylor's diphenylborinic acid catalyst¹⁹ was achieved with

Scheme 3. Catalyst Recovery and Recycling in the Asymmetric Alcohol C–H Allylation of 7^a

^aYields are of material isolated by silica gel chromatography. The third and higher cycles provided yields ranging between 50 and 60%. See Supporting Information for further details.

excellent levels of site-selectivity to provide *mono*-benzyl ether *mono*-Boc-carbonate 11. Modified Bartlett halocyclization of the Boc-carbonate²⁰ followed by treatment of the crude mixture with K₂CO₃ in methanol delivered the epoxy diol 12 in 48% yield as an 8:1 mixture of diastereomers. Protection of the diol as the acetonide provides the requisite epoxide 13 (Scheme 4).

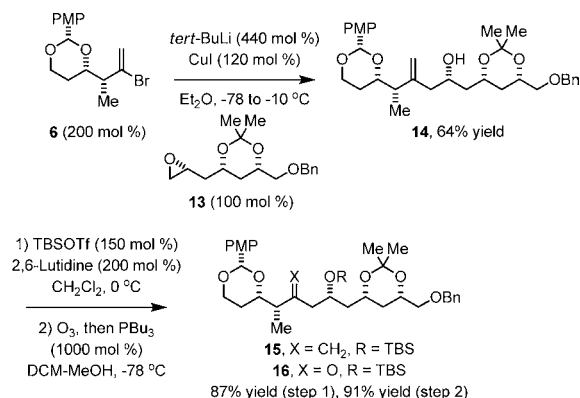
Scheme 4. Conversion of Homoallylic Alcohol 8 to Epoxide 13^a

^aYields are of material isolated by silica gel chromatography. See Supporting Information for further details.

To form the C9–C20 substructure of tetrafibricin, the cuprate derived from alkenyl bromide 6 was reacted with epoxide 13.^{21,22} Initially, it was desired to use the commercially available lithium 2-thienyl(cyano)cuprate solution to form the higher order mixed organocuprate reagent,^{21b,22b} so as to limit the stoichiometry of alkenyl bromide 6 used in the transformation. However, this method failed to deliver serviceable quantities of adduct 14. In contrast, using cuprous iodide as the copper source with 2 equiv of alkenyl bromide, the product of epoxide opening was obtained in good yield, based on 13.^{21a,22a} Silyl protection of alcohol 14, followed by ozonolysis of olefin 15, delivered the C9–C20 fragment of tetrafibricin in 10 linear steps from the commercial malic acid derivative 7 and in 15 total steps (Scheme 5).

In summary, we report a synthetic route to the C9–C20 substructure of tetrafibricin using ruthenium catalyzed enantioselective *syn*-crotylation to construct C9–C13 and iridium catalyzed asymmetric alcohol C–H allylation to construct C14–C20. This synthetic exercise led to the observation that the cyclometalated iridium complex used in the C–H allylation of primary alcohols may be recovered and recycled. The ability to perform multiple rounds of catalyst recovery and recycling, along with the ability to engage alcohols directly as partners for catalytic C–C coupling in the absence of

Scheme 5. Reductive Coupling of Alkenyl Bromide **6** and Epoxide **13** To Form the C9–C20 Substructure of Tetrafibricin^a



^aYields are of material isolated by silica gel chromatography. See Supporting Information for further details.

discrete alcohol-to-aldehyde redox reactions, makes the present iridium catalyzed allylation protocol both cost-effective and step-economical. A more systematic study of catalyst recovery and recycling, including immobilization of the catalyst on a solid support, will be reported in due course.

■ ASSOCIATED CONTENT

Supporting Information

Spectral data for all new compounds (¹H NMR, ¹³C NMR, IR, HRMS). This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: mkrische@mail.utexas.edu.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

Acknowledgment is made to the Robert A. Welch Foundation (F-0038), the NIH-NIGMS (RO1-GM093905), and the Center for Green Chemistry and Catalysis for partial support of this research. Dr. Taichiro Touge and Dr. Hideo Shimizu of Takasago are thanked for the generous donation of SEGPHOS ligands.

■ REFERENCES

- (1) Isolation and structure assignment: (a) Kamiyama, T.; Umino, T.; Fujisaki, N.; Satoh, T.; Yamashita, Y.; Ohshima, S.; Watanabe, J.; Yokose, K. *J. Antibiot.* **1993**, *46*, 1039. (b) Kamiyama, T.; Itezono, Y.; Umino, T.; Satoh, T.; Nakayama, N.; Yokose, K. *J. Antibiot.* **1993**, *46*, 1047. (c) Kobayashi, Y.; Czechtizky, W.; Kishi, Y. *Org. Lett.* **2003**, *5*, 93.
- (2) Satoh, T.; Yamashita, Y.; Kamiyama, T.; Watanabe, J.; Steiner, B.; Hadvary, B.; Arisawa, M. *Thromb. Res.* **1993**, *72*, 389.
- (3) Satoh, T.; Yamashita, Y.; Kamiyama, T.; Arisawa, M. *Thromb. Res.* **1993**, *72*, 401.
- (4) Pawlak, J.; Nakanishi, K.; Iwashita, T.; Borowski, E. *J. Org. Chem.* **1987**, *52*, 2896.
- (5) (a) Satoh, T.; Kouns, W. C.; Yamashita, Y.; Kamiyama, T.; Steiner, B. *Biochem. J.* **1994**, *301*, 785. (b) Satoh, T.; Kouns, W. C.;

Yamashita, Y.; Kamiyama, T.; Steiner, B. *Biochem. Biophys. Res. Commun.* **1994**, *204*, 325.

- (6) BouzBouz, S.; Cossy, J. *Org. Lett.* **2004**, *6*, 3469.
- (7) (a) Gudipati, V.; Bajpai, R.; Curran, D. P. *Collect. Czech. Chem. Commun.* **2009**, *74*, 774. (b) Zhang, K.; Gudipati, V.; Curran, D. P. *Synlett* **2010**, 667. (c) Gudipati, V.; Curran, D. P. *Tetrahedron Lett.* **2011**, *52*, 2254. (d) Doctoral thesis: <http://d-scholarship.pitt.edu/6572/>.
- (8) Friestad, G. K.; Sreenilayam, G. *Org. Lett.* **2010**, *12*, 5016.
- (9) (a) Lira, R.; Roush, W. R. *Org. Lett.* **2007**, *9*, 533. (b) Kister, J.; Nuhant, P.; Lira, R.; Sorg, A.; Roush, W. R. *Org. Lett.* **2011**, *13*, 1868. (c) Nuhant, P.; Roush, W. R. *J. Am. Chem. Soc.* **2013**, *135*, 5340.
- (10) Kumpulainen, E. T. T.; Kang, B.; Krische, M. J. *Org. Lett.* **2011**, *13*, 2484.
- (11) For recent reviews on C–C bond forming hydrogenation and transfer hydrogenation, see: (a) Bower, J. F.; Krische, M. J. *Top. Organomet. Chem.* **2011**, *43*, 107. (b) Hassan, A.; Krische, M. J. *Org. Process Res. Dev.* **2011**, *15*, 1236. (c) Moran, J.; Krische, M. J. *Pure Appl. Chem.* **2012**, *84*, 1729. (d) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Gao, X.; Itoh, T.; Krische, M. J. *Nat. Prod. Rep.* **2014**, DOI: 10.1039/C3NP70076C.
- (12) (a) Gao, X.; Woo, S. K.; Krische, M. J. *J. Am. Chem. Soc.* **2013**, *135*, 4223. (b) Lu, Y.; Woo, S. K.; Krische, M. J. *J. Am. Chem. Soc.* **2011**, *133*, 13876. (c) Del Valle, D. J.; Krische, M. J. *J. Am. Chem. Soc.* **2013**, *135*, 10986. (d) Waldeck, A. R.; Krische, M. J. *Angew. Chem., Int. Ed.* **2013**, *52*, 4470. (e) Han, S. B.; Hassan, A.; Kim, I. S.; Krische, M. J. *J. Am. Chem. Soc.* **2010**, *132*, 15559.
- (13) For an overview of “redox economy” in organic synthesis, see: Baran, P. S.; Hoffmann, R. W.; Burns, N. Z. *Angew. Chem., Int. Ed.* **2009**, *48*, 2854.
- (14) For enantioselective iridium catalyzed alcohol C–H allylations employing allyl acetate, see: (a) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. *J. Am. Chem. Soc.* **2008**, *130*, 6340. (b) Kim, I. S.; Ngai, M.-Y.; Krische, M. J. *J. Am. Chem. Soc.* **2008**, *130*, 14891. (c) Lu, Y.; Kim, I. S.; Hassan, A.; Del Valle, D. J.; Krische, M. J. *Angew. Chem., Int. Ed.* **2009**, *48*, 5018. (d) Hassan, A.; Lu, Y.; Krische, M. J. *Org. Lett.* **2009**, *11*, 3112. (e) Schmitt, D. C.; Dechert-Schmitt, A.-M. R.; Krische, M. J. *Org. Lett.* **2012**, *14*, 6302. (f) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Krische, M. J. *Angew. Chem., Int. Ed.* **2013**, *52*, 3195.
- (15) For enantioselective ruthenium catalyzed alcohol C–H crotylation via diene hydrohydroxyalkylation, see: (a) Zbieg, J. R.; Moran, J.; Krische, M. J. *J. Am. Chem. Soc.* **2011**, *133*, 10582. (b) Zbieg, J. R.; Yamaguchi, E.; McInturff, E. L.; Krische, M. J. *Science* **2012**, *336*, 324. (c) McInturff, E. L.; Yamaguchi, E.; Krische, M. J. *J. Am. Chem. Soc.* **2012**, *134*, 20628.
- (16) Oikawa, Y.; Nishi, T.; Yonemitsu, O. *Tetrahedron Lett.* **1983**, *24*, 4037.
- (17) (a) Miller, R. B.; Reichenbach, T. *Tetrahedron Lett.* **1974**, *15*, 543. (b) Miller, R. B.; McGarvey, G. J. *Org. Chem.* **1978**, *43*, 4424. (c) Miller, R. B.; McGarvey, G. J. *Org. Chem.* **1979**, *44*, 4623.
- (18) Gao, X.; Townsend, I. A.; Krische, M. J. *J. Org. Chem.* **2011**, *76*, 2350.
- (19) Lee, D.; Williamson, C. L.; Chan, L.; Taylor, M. S. *J. Am. Chem. Soc.* **2012**, *134*, 8260.
- (20) (a) Bartlett, P. A.; Meadows, J. D.; Brown, E. G.; Morimoto, A.; Jernstedt, K. K. *J. Org. Chem.* **1982**, *47*, 4013. (b) For recent examples: Smith, A. B., III; Lin, Q.; Nakayama, K.; Boldi, A. M.; Brook, C. S.; McBriar, M. D.; Moser, W. H.; Sobukawa, M.; Zhuang, L. *Tetrahedron Lett.* **1997**, *38*, 8675. (c) Stivala, C. E.; Gu, Z.; Smith, L. L.; Zakarian, A. *Org. Lett.* **2012**, *14*, 804.
- (21) For seminal reports of epoxide ring openings employing alkenyl cuprate reagents and their thienyl modified analogues, respectively, see: (a) Herr, R. W.; Wieland, D. M.; Johnson, C. R. *J. Am. Chem. Soc.* **1970**, *92*, 3813. (b) Lipshutz, B. H.; Koerner, M.; Parker, D. A. *Tetrahedron Lett.* **1987**, *28*, 945.
- (22) For selected examples from the total synthesis literature of related epoxide ring openings employing alkenyl cuprate reagents and their thienyl modified analogues, respectively, see: (a) Guo, J.; Duffy, K. J.; Stevens, K. L.; Dalko, P. I.; Roth, R. M.; Hayward, M. M.; Kishi,

Y. *Angew. Chem., Int. Ed.* **1998**, 37, 187. (b) Smith, A. B., III; Friestad, G. K.; Barbosa, J.; Bertounesque, E.; Hull, K. G.; Iwashima, M.; Qiu, Y.; Salvatore, B. A.; Spoors, P. G.; Duan, J. J.-W. *J. Am. Chem. Soc.* **1999**, 121, 10468.

■ **NOTE ADDED AFTER ASAP PUBLICATION**

An updated version was reposted January 14, 2014 which includes reference 11d. On January 15, 2014 an error in Scheme 3 was corrected.